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Figure 20-29 shows a reversible cycle through which 1.00 mol of a monatomic ideal gas is taken. Volume  $V_c = 8.00V_b$ . Process bc is an adiabatic expansion, with  $p_b = 10.0 \text{ atm}$  and  $V_b = 1.00 \times 10^{-3} \text{ m}^3$ . For the cycle, find (a) the energy added to the gas as heat, (b) the energy leaving the gas as heat, (c) the net work

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