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## **All Chemistry Formulas**

Molecular formulas are more limiting than chemical names and structural formulas. Empirical and Molecular Formulas. The simplest types of chemical formulas are called empirical

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formulas, which indicate the ratio of each element in the molecule. The empirical formula is the simplest whole number ratio of all the atoms in a molecule. For example:

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Molecular formulas are derived by

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comparing the compound's molecular or molar mass to its empirical formula mass. As the name suggests, an empirical formula mass is the sum of the average atomic masses of all the atoms represented in an empirical formula.

## **3.2 Determining Empirical and Molecular Formulas ...**

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Molecular formulas are derived by comparing the compound's molecular or molar mass to its empirical formula mass. As the name suggests, an empirical formula mass is the sum of the average atomic masses of all the atoms represented in an empirical formula.

## **3.2 Determining Empirical and**

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